

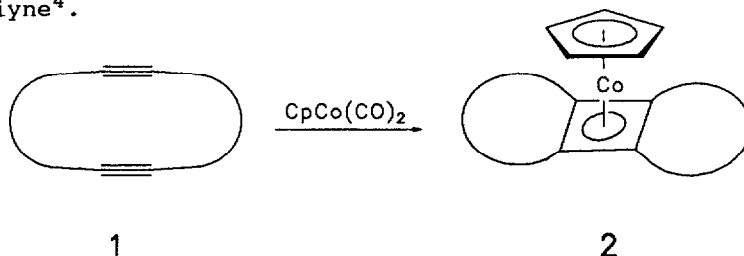
A ONE STEP SYNTHESIS OF A PENTAMETHYLENE-BRIDGED SUPERPHANE OF A CpCo-
STABILIZED CYCLOBUTADIENE COMPLEX¹

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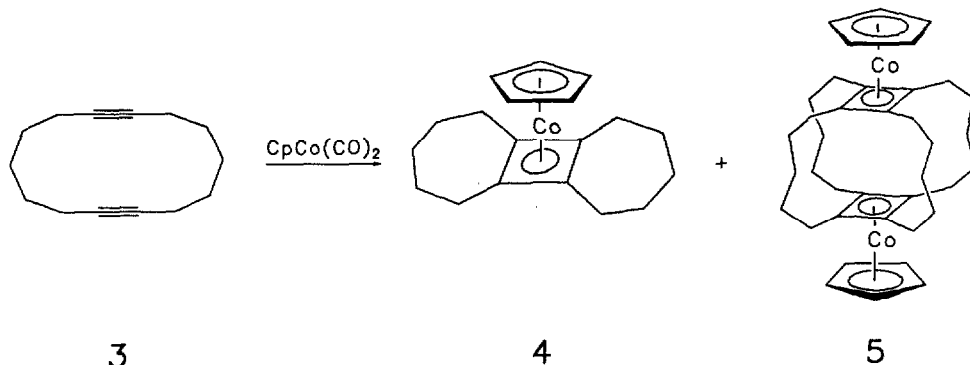
Heating 1,8-cyclotetradecadiyne with dicarbonyl(η^5 -cyclopentadienyl)cobalt($\text{CpCo}(\text{CO})_2$) gives in 2-7% yield the [5₄]-
(1,2,3,4)cyclobutadienophane complexed with two (C_5H_5)Co units
in one step.

For the thermal reaction of cyclic nonconjugated diynes **1** with dicarbonyl(η^5 -cyclopentadienyl)cobalt ($\text{CpCo}(\text{CO})_2$) or the corresponding complex with 1,4-cyclooctadiene (cod), $\text{CpCo}(\text{cod})$, the intramolecular reaction yielding a tricyclic cyclobutadiene complex **2** is the rule²⁻⁴. Exceptions are found in the case of 1,6-cyclodecadiyne³ and 1,5-cycloundecadiyne⁴.



A comparison of the yields obtained for the intramolecular complexes of 1,6-cyclodecadiyne (6%)³, 1,6-cycloundecadiyne (32%)⁴, 1,7-cyclododecadiyne (86%)² and 1,7-cyclotridecadiyne (65%)¹ suggests that the less strained cyclodiyne should give good yields of the intramolecular products **2**. The low yields reported by King et al.² for 1,8-

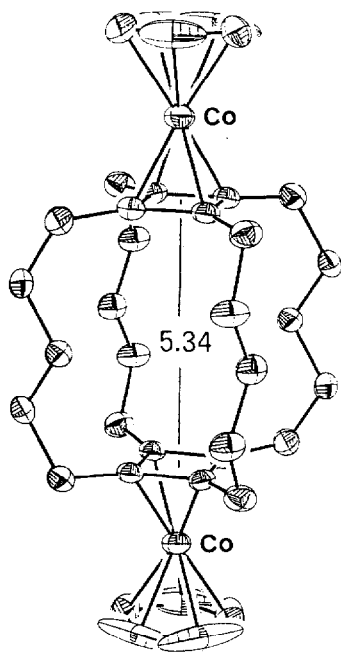
cyclotetradecadiyne(3) (2-10% yield)⁵ led us to reinvestigate the reaction of 3 with $\text{CpCo}(\text{CO})_2$. We are not only able to confirm the



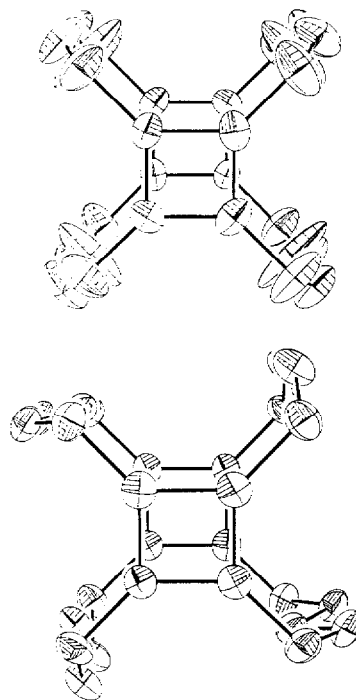
relatively low yield reported for 4 in *n*-octane (16-18% with $\text{CpCo}(\text{CO})_2$, 29% with $\text{CpCo}(\text{cod})$), but we were also able to isolate the intermolecular cycloaddition product 5, the [5₄](1,2,3,4)cyclobutadienophane complexed with two CpCo units in 2-7% yield. By carrying out the reaction in decalin at 180°, we obtained 4 in 70% yield, while the yield of 5 remains on the order of 5-7%. To our knowledge this is the first CpCo-capped cyclobutadiene superphane with chains which are longer than three carbon atoms. So far, superphanes of metal complexed cyclobutadiene have been reported with trimethylene chains containing two CpCo-units, one CpCo and one Cp*Co unit or one $\text{Fe}(\text{CO})_3$ unit, respectively^{3,6}.

The structural assignment of 5 reveals two crystallographically independent molecules⁷ in the unit cell whose long axes are inclined to each other by 75.3°. The pentamethylene bridges in both molecules adopt a zig-zag arrangement (Fig.1a) leading to a bend pin-wheel conformation. In one case this conformation is more rigid than in the other (Fig.1b). The distances between the cyclobutadiene units in 5 are found to be 5.34 Å and 5.37 Å, respectively. A similar zig-zag conformation has been found in 7,10,18,21-tetramethoxy[5.5]paracyclophane where both benzene rings are separated by 5.1 Å⁸.

a



b

**Figure 1**

a) Side view of the molecular structure of **5**.

b) Thermal ellipsoids (40% probability) of the cyclobutadienophane skeletons of the two independent molecules of **5** in the unit cell.

In both cases the hydrogens are omitted for clarity.

Table 1 Most relevant spectroscopic data of **4** and **5**.

4: $^1\text{H-NMR}$ (CDCl_3 , 300 MHz): $\delta=4.8$ (s,5H), 2.1-1.9 (m,10H), 1.7-1.5 (m,8H), 1.0-0.8 (m,2H); $^{13}\text{C-NMR}$ (CDCl_3 , 300 MHz): $\delta=79.6$, 78.8, 30.5, 27.9, 26.7.

5: $^1\text{H-NMR}$ (CDCl_3 , 300 MHz): $\delta=4.4$ (s,10H), 2.28 (s,br,16H), 1.27 (s,br,24H); $^{13}\text{C-NMR}$ (CDCl_3 ,200 MHz): $\delta=80.7$, 80.0, 30.0, 27.2, 26.3; IR (CDCl_3): $\nu[\text{cm}^{-1}]=2924$, 2848, 1429, 1310, 1261; UV/VIS (pentane): $\lambda_{\text{max}}[\text{nm}](\lg \epsilon)=197(4.27)$, 225 sh(4.13), 266(4.43), 380(2.70).HRMS: calc. 624.2564; found 624.2576.

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References

- 1 Dedicated to Professor Klaus Weinges on the occasion of his 65th birthday.
- 2 King, R.B.; Efraty, A., *J.Am.Chem.Soc.* **1970**, *92*, 6071-6073; King, R.B.; Efraty, A., *J.Am.Chem.Soc.* **1972**, *92*, 3021-3025.
- 3 Gleiter, R.; Karcher, M.; Ziegler, M.L.; Nuber, B., *Tetrahedron Lett.* **1987**, *28*, 195-198; Gleiter, R.; Kratz, D., *Tetrahedron Lett.* **1990**, *31*, 5893-5896.
- 4 Gleiter, R.; Kratz, D.; Ziegler, M.L.; Nuber, B., *Tetrahedron Lett.* **1990**, *31*, 6175-6178.
- 5 With $\text{CpCo}(\text{CO})_2$ and $\text{CpCo}(\text{cod})$ the yields of **4** reported are 2-4% and 10%, respectively².
- 6 Gleiter, R.; Schehlmann, V. *Angew.Chem.* **1990**, *102*, 1450-1452; *Angew.Chem.Int.Ed.Engl.* **1990**, *29*, 1426-1427.
- 7 Unit cell constants for **5**: $a=8.903(5)\text{\AA}$, $b=9.520(6)\text{\AA}$, $c=20.98(2)\text{\AA}$, $\alpha=99.78(6)$, $\beta=92.98(6)$, $\gamma=113.57(4)$, $V=1591.9\text{\AA}^3$, $Z=2$. Space group $\text{P}\bar{1}$. Data were collected on a SYNTEX R3 diffractometer using a crystal of dimensions $0.04\times 0.35\times 0.5\text{ mm}^3$. Ω -scan ($3.0<2\theta<52.5^\circ$), $\text{MoK}\alpha$ radiation; 5973 reflexions of which 3290 were judged to be independent ($I>2.5\sigma(I)$). All crystallographic calculations used the SHELXTL PLUS program package (January 1989). Empirical absorption corrections (6 reflexions, $8.0<2\theta<40^\circ$). Relative range of transmissions factors 0.84-1.00 ($\mu=11\text{ cm}^{-1}$). Least squares refinement with ($w=1/\sigma^2(F)$). The structure was solved by direct methods (TREF, SHELXTL PLUS). The hydrogen atoms were located by option HFIX of the SHELXTL PLUS program; $R=0.064$, $R_w=0.055$. Complete structural data have been submitted to the **Cambridge University Crystallographic Center**.
- 8 Staab, H.A.; Starker, B.; Krieger, C. *Chem.Ber.* **1983**, *116*, 3831-3845.